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# SPECTROSCOPIC AND LASER PROPERTIES OF LinbO<sub>3</sub>: Dy<sup>3+</sup> CRYSTALS

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The spectroscopic properties of trivalent dysprosium (Dy³+) doped LiNbO₃ crystals have been investigated at various temperatures. Absorption, emission, *excitation* and lifetime measurements have been performed and discussed in the framework of Judd—Ofelt approach. The stimulated emission cross sections of the strongest transitions of Dy³+ ion have been estimated. A stimulated emission has been demonstrated in the near infrared.

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# 1. Introduction

Rare-earth doped lithium niobate, owing to good electro-optic, acousto-optic and nonlinear properties of the host matrix, offers technologically important possibilities for planar waveguide lasers and amplifiers and received considerable attention in recent years.

Laser action in Nd<sup>3+</sup>, Ho<sup>3+</sup> and Tm<sup>3+</sup> doped LiNbO<sub>3</sub> (LNB) has been reported [1] and spectroscopic properties of these crystals as well as Er<sup>3+</sup> [2] and Cr<sup>3+</sup> [3] doped materials have been studied. Diode pumped channel waveguide lasers have been realized in Nd<sup>3+</sup> [4], Yb<sup>3+</sup> [5] and Er<sup>3+</sup> [6] activated LiNbO<sub>3</sub>.

Recent reports indicate the interest of activating the optical glasses by  $Dy^{3+}$  ions for the purposes of 1.3  $\mu$ m band fiber telecommunication [7] and dysprosium is being compared with the most widely used dopant at present — praseodymium [8]. However, literature on the spectroscopy of  $Dy^{3+}$  doped materials is not exhaustive and, to our knowledge, laser action in  $Dy^{3+}$  activated solids has been only observed at the 3  $\mu$ m band [9] in  $BaY_2F_8$  crystal.

The aim of this work was to examine a new optically active material which may be of considerable technological significance. We report on the spectroscopic investigation of  $\mathrm{Dy}^{3+}$  ions in  $\mathrm{LiNbO_3}$ . Various absorption, emission and excitation spectra have been investigated for several activator concentrations at temperatures between 10 and 300 K. Low temperature spectroscopic data were used to establish  $\mathrm{Dy}^{3+}$  energy level diagram in  $\mathrm{LiNbO_3}$  and to investigate the inhomogeneous character of optical transitions. The Judd—Ofelt theory has been applied to the optical absorption linestrengths measured at room temperature to determine  $\Omega_i$  intensity parameters, which were then used to predict the radiative decay rates and branching ratios for transitions from the metastable  ${}^4F_{912}$  state of  $\mathrm{Dy}^{3+}$  ion. These results were compared with those for other known dysprosium systems.

# 2. Experimental methods

LiNbO<sub>3</sub>: Dy<sup>3+</sup> samples with concentrations 0.31 and 1.7 at.% were grown parallel to the *(c)* direction using the Czochralski technique at I.T.M.E. Laboratory in Warsaw. Fluorescence and excitation spectra were obtained using a tunable laser operating with Coumarine 460-481 dyes, pumped by a nitrogen laser (10 ns pulse length, 10 Hz repetition rate and 10 mJ energy per pulse at 337 nm). CW excitation in the blue region was achieved by a ILA-120 3 W argon ion laser. The spectra were recorded using a UDM-1000 rnonochromator with dispersion of 11 cm<sup>-1</sup>/mm and detected by RCA C-31034-02 cooled AsGa photomultiplier. Data acquisition was obtained using a SR 400 photon counting system controlled with a PC computer. Fluorescence lifetime measurements were made using a Tennelec/Nucleus MCS multichannel analyzer. Sample cooling was provided by a closed circuit He optical cryostat allowing to vary temperature between 10 and 300 K.

#### 3. Results and discussion

# 3.1. Absorption and emission spectra

Room temperature absorption spectrum, for the light propagating parallel to the c axis, in the 400—650 nm range of 1.7% Dy<sup>3+</sup> doped LiNbO<sub>3</sub> is presented in Fig. 1. It shows several, characteristic for the  $4f^9 \rightarrow 4f^9$  transitions of trivalent dysprosium, groups of lines which correspond to transitions from the  $^6H_{15/2}$ 

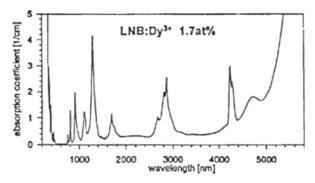


Fig. 1. Absorption spectrum of 1.7 at.% Dy<sup>3~</sup>: LiNbO<sub>3</sub> crystal at room temperature.

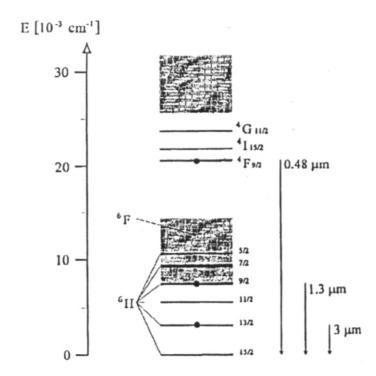


Fig. 2. Energy level scheme of an isolated  $Dy^{3+}$  ion. Emission transitions reported in literature are indicated by arrows [1, 7, 8, 16].

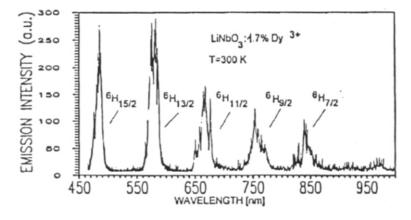


Fig. 3. Room temperature visible emission spectrum of Dy<sup>3+</sup> in LiNbO<sub>3</sub> crystal.

ground state to the different multiplets, see also energy level scheme in Fig. 2. Below about 500 nm the background absorption of the host crystal begins to increase and it grows rapidly at about 250 nm. After excitation by the 476 nm blue line of an argon laser, emission of the crystals was recorded between 300 nm and 2  $\mu$ m. Figure 3 illustrates the room temperature emission spectrum of 1.7% Dy<sup>3+</sup>: LiNbO<sub>3</sub> in the visible range, no infrared (IR) emission was detected. It was concluded that in LiNbO<sub>3</sub>: Dy<sup>3+</sup> visible fluorescence is due to transitions from the excited  ${}^4F_{912}$  state to the lower  ${}^6H$ ,  $F_J$  states and, though these transitions are forbidden by the spin selection rule, they were observed as relatively intense.

It could be seen that room temperature lines are too broad for precise as-

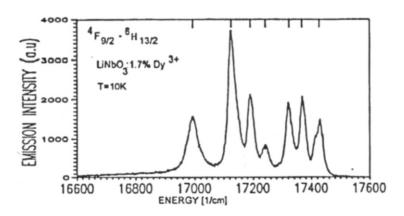


Fig. 4. Part of the visible emission spectrum of Dy<sup>3</sup>+ in LiNbO<sub>3</sub> crystal measured at 10 K.

TABLE I

Energy levels of Dy<sup>3+</sup> ion in LiNbO<sub>3</sub> measured at 10 K.

[SLJ] manifold	$E [cm^1]$
6111	0 14 45 112 222 2 02 422 (02
$^{6}H1_{5/2}$	0, 14, 45, 113, 222,3 03, 422, 603
$^{6}\text{H}_{13/2}$	3535, 3595, 3643, 3725, 3775, 3840, 3969
$^{6}H_{11/2}$	5878, 5928, 5991, 6059, 6117, 6221
$^{6}H_{9/2}+^{6}F_{11/2}$	7678, 7700, 7749, 7798*, 7815, 7843, 7891, 7937, 8004, 8093*
$^{6}H_{7/2} + ^{6}F_{9/2}$	90621, 91655, 92060, 93526, 9838, 9853, 9878, 9943, 9979, 10035
$^{6}H_{5/2}$	<del>_</del>
$^{6}F_{7/2}$	11080, 11108, 11143*, 11176*
$^{6}F_{5/2}$	12445, 12487, 12555
${}^{6}F_{3/2}$	13271, 13318*
$^{6}F_{1/2}$	<del>_</del>
$^{4}F_{9/2}$	20963, 20974*, 21069, 21097, 21113
$^{4}I_{l5/2}$	21945, 21953, 21978, 22009, 22211, 22216, 22255

<sup>\*</sup> denotes less accurate results.

signment. The determination of individual Stark levels of the  $Dy^{3+}$  ions was made by analyzing various low temperature absorption, excitation and dye laser excited emission spectra which showed sharper lines, see Fig. 4. The linewidth of the lowest Stark component of the  $^4F9/2$  metastable state in 0.3%  $Dy^{3+}$  crystal was measured to be 4 cm<sup>-1</sup> at 10 K, indicating a smaller amount of the inhomogeneous broadening than reported for  $Er^{3+}$  ion in LiNbO<sub>3</sub> [10]. A summary of the energy levels derived from our measurements is given in Table 1.

# 3.2. Optical transition intensity analysis

Once the energy levels have been determined, the Judd—Ofelt [11, 12] intensity analysis was performed using the  $\pi$  and  $\sigma$  polarized absorption spectra as

proposed in Ref. [13]. Since the details of the Judd—Ofelt theory, its precision and drawbacks have been extensively analyzed elsewhere [11—13] we have presented here only the essential results. The central result of the Judd—Ofelt theory is that the oscillator strength  $f_{calc}$  of an electric dipole transition between rare-earth ion multiplets  $(J \rightarrow J')$  can be expressed by

TABLE II Measured and calculated oscillator strengths for Dy<sup>3+</sup> ion in LiNLO<sub>3</sub>.

[S'L'J'] manifold	λ[nrn]	$f_{exp} [10^{-}6]$	$f_{calc}$	
$^{6}H_{13/2}$	2869	3.94	2.48	
$^{6}H_{11/2}$	1690	2.25	2.12	
$^{6}H_{9/2} +  ^{6}F_{11/2}$	1290	13.06	13.12	
$^6H_{7/2} + ^6F_{9/2}$	1100	2.49	4.34	
$^6F_{7/2}$	910	3.28	3.37	
$^6F_{5/2}$	806	1.56	1.52	
$^{6}F_{3/2}$	757	0.290	0.287	
$^4F_{9/2}$	475	0.241	0.265	
$^{4}I_{l5/2}$	453	0.830	0.773	
$^4G_{11/2}$	428	0.417	0.15	
$^{4}F_{712} + ^{4}I_{1312} + ^{4}M_{21/2}$	390	3.96	3.78	
$^4M_{19/2} + ^4P_{5/2}$	372	3.70	2.48	
$^{4}I_{11/2} + ^{4}P_{7/2}$	358	6.71	6.26	

RMS dev. = 2.24 x 10<sup>-7</sup>, 
$$\Omega_2$$
 = 9.57 x 10<sup>-20</sup> cm<sup>2</sup>,  
 $\Omega_4$  = 2.63 x 10<sup>-20</sup> cm<sup>2</sup>,  $\Omega_6$  = 2.52 x 10<sup>-20</sup> cm<sup>2</sup>.  

$$f_{calc}(aJ,bJ') = \frac{8\pi^2 mv}{3h(2J+1)n^2} \chi \sum_{t=2,4,6} \Omega_t |\langle 4f^n | a,J || U^{(t)} || 4f^n | b,J' |\rangle|^2, \quad (1)$$

where h is Planck's constant, J is the angular momentum of the initial level,  $x = n(n^2 + 2)^{2/9}$  is a local field correction factor,  $\langle a \parallel U^{(l)} \parallel b \rangle$  are the doubly reduced matrix elements and  $\Omega$  are empirically determined parameters. The experimental oscillator strength  $f_{exp}$  for an absorption transition is defined as

$$f_{\rm exp} = \frac{mc^2}{\pi e^2} \int \sigma(v) dv \tag{2}$$

where m and e are the electron mass and charge, respectively, e is the light velocity, e is the optical frequency and  $\sigma(v)$  is the absorption cross section.

From the least square fit of measured  $f_{exp}$  and calculated  $f_{calc}$  oscillator strengths the three intensity  $\Omega_t$  parameters were evaluated. In performing tile calculations the reduced matrix elements were taken from Carnall et al. [14] and the index of refraction n as a function of wavelength was taken from Ref. [15]. Table II shows tile average wavelengths for the analysed transitions together with

TABLE III Predicted radiative transition probabilities A and branching ratios  $\beta$  from the  ${}^4F_{9/2}$  level of Dy<sup>3+</sup> ion in LiNbO<sub>3</sub>,  $\tau_{\rm R} = 298 \,\mu \rm s$ .

Final state	λ	A	$oldsymbol{eta_{ m calc}}$	$oldsymbol{eta_{ ext{means}}}$
[S'L'J']	[nm]	$[s^{-1}]$	[%]	[%]
$^{6}H_{15/2}$	475	557	17	20
$^{6}H_{13/2}$	569	2337	69	46
$^{6}H_{11/2}$	661	228	7	17
$^{6}H_{9/2} + ^{6}F_{11/2}$	752	118	3.7	9
$^{6}H_{7/2} + ^{6}F_{9/2}$	836	72	2	8
$^{6}H_{5/2} + ^{6}F_{7/2}$	1014	19	0.6	0
$^{6}F_{5/2}$	1180	24	0.7	0
${}^{6}F_{3/2}$	1305	0	0	0

the measured  $f_{exp}$  and calculated  $f_{calc}$  oscillator strengths for all measured absorption transitions. The resulting set of Judd—Ofelt, parameters was found to be:  $\Omega=9.57 \times 10^{-20} \, \mathrm{cm^2}$ ,  $\Omega_4=2.63 \times 10^{-20} \, \mathrm{cm^2}$  and  $\Omega_6=2.52 \times 10^{-20} \, \mathrm{cm^2}$ . The fit between the measured and calculated oscillator strengths values, shown in Table II, was good as indicated by the root mean square (RMS) deviation of 2.24 x  $10^{-7}$  which is comparable to values found by applying Judd—Ofelt theory to  $\mathrm{Dy^{3^+}}$  ion in other systems [7,16]. It could be noticed that LNB presents oscillator strengths for the absorption transitions very similar to those of oxide glasses [16]. From the calculated set of  $\Omega_t$  intensity parameters the electric dipole transition probabilities A(aJ, bJ') for emission between J manifolds of  $\mathrm{Dy^{3^+}}$  were calculated using the following equation:

$$A(aJ,bJ') = \frac{64\pi^4 e^2 v^3}{3h(2J+1)c^3} \chi \sum_{t=2,4,6} \Omega_t |< 4f^n | a,J || U^{(t)} || 4f^n | b,J' |> |^2.$$
 (3)

Calculated radiative transition probabilities A from Cite excited  ${}^4F_{9/2}$  state together with the resulting branching ratios  $f_{calc}$  are given in Table III. Six emission bands were observed due to the transitions from the  ${}^4F_{9/2}$  level to lower excited states, see Fig. 3. The corresponding branching ratios were obtained from the nonpolarized fluorescence spectra by comparing relative areas under the emission peaks and are listed in the last column of Table III. It could be seen that the highest calculated value of the branching ratio is for the  ${}^4F_{9/2} \rightarrow {}^6H_{13/2}$  transition which is found to be in agreement with the experimentally determined  $\beta_{\text{means}}$  value. The largest discrepancy between experiment and Chicory is for the  ${}^4F_{9/2} \rightarrow {}^6H_{15/2}$  transition, where the measured branching ratio is substantially longer than the calculated value. The radiative lifetime of the  ${}^4F_{9/2}$  state was calculated to be 298  $\mu$ s and compares reasonably with the radiative lifetime value of 371  $\mu$ s found for oxide glass [16]. However, radiative lifetime calculated for the fluorozincate glass was much longer, of 1.26 ms [17].

# 3.3. Fluorescence dynamics and time-resolved experiments

Figure 5 shows the decay profiles of the  ${}^4F_{9/2}$  level. For all investigated Dy  ${}^{3+}$  concentrations these decays are slightly nonexponential suggesting that the concentration quenching could be active for concentration as low as 0.3% of dysprosium. From the long time part of the decay the fluorescence lifetime was determined to be of 245  $\mu$ s. The time behavior of the  ${}^4F_{9/2}$  fluorescence in LiNbO3 was studied at different temperatures and was found to be nearly temperature and activator concentration independent between 10 and 300 K suggesting its predominantly radiative character. This is consistent with the large energy gap of 7680 cm<sup>-1</sup> to the next lower lying level. Difference between calculated radiative and experimental lifetimes of Dy  ${}^{3+}$  could be due to the overestimation of the actual concentration of activator in the investigated sample. It is also possible that the  ${}^4F_{9/2}$  emission is affected by the background absorption of the LiNbO3 host.

Time-resolved (TR) excitation spectroscopy has been used for further studies of the character of line broadening. The dye laser was scanned across the  ${}^6H_{15/2} \rightarrow {}^4F_{9/2}$  absorption and the total  ${}^4F_{9/2}$  fluorescence was registered at various delays after the excitation pulse. From comparison of the excitation spectra presented in Fig. 6 no linewidth or lineshape variation with the delay time could be observed which suggests the same  $\Pr^{3+}$  ions in SrLaGa<sub>3</sub>O<sub>7</sub> lattice; longer lived regular ions and ions in minority sites characterized by much stronger coupling to the lattice and efficient nonradiative decay.

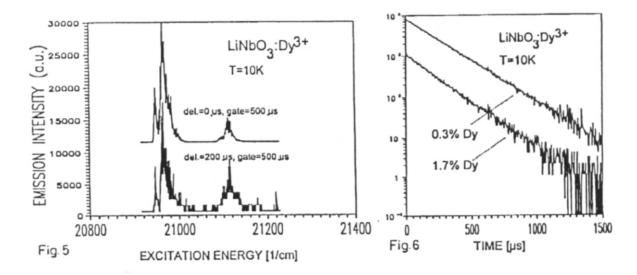


Fig. 5. Fluorescence decays at 10 K in LiNbO<sub>3</sub> crystals with 0.3 and 1.7 at.% of Dy<sup>3+</sup>. Fig. 6. Time-resolved excitation s1)ectra. of the  ${}^6H_{15/2} \rightarrow {}^4F_{9/2}$  transition in Dy<sup>3+</sup>: LiNbO<sub>3</sub> at 10 K.

When comparing various dysprosium doped media [12,13,16] it is apparent that the emission properties of LiNbO<sub>3</sub> are more influenced by nonradiative processes.

By comparing the polarized resonant absorption and emission spectra the emission cross section  $\sigma_{em}$  of the visible emissions has been determined. The strongest fluorescence peaks at 485 nm and 580 nm are characterized by

 $\sigma_{em}(485 \text{ nm}) = 1.6 \text{ x } 10^{-21} \text{ cm}^2 \text{ and } \sigma_{em}(580 \text{ nm}) = 7.8 \text{ x } 10^{-21} \text{ cm}^2 \text{ value}$ . The emission cross section could be also calculated from the absolute transition probability and the fluorescence linewidth

$$\sigma_{em} = \frac{\lambda^2 \beta}{8\pi c n^2 \tau_R \Delta v_{eff}},\tag{4}$$

TABLE IV

where  $\lambda$  is the wavelength of the fluorescence peak,  $\Delta v_{eff}$  is the effective fluorescence linewidth and  $\beta$  and  $\tau_R$  are branching ratio and radiative lifetime, respectively. Using the measured values (see Fig. 1 and Table I) and the results of Judd—Ofelt calculations (see Table III) the emission cross sections were found to be  $\sigma_{em} = 2.2 \times 10^{-21} \text{ cm}^2$  and  $\sigma_{em} = 12 \times 10^{-21} \text{ cm}^2$ , respectively for 485 nm and 580 nm transitions. It is seen that the blue emission cross section value calculated from the emission spectra is lower than the theoretical one which may be due to reabsorption and excited state absorption losses.

The spectroscopic properties of various Dy<sup>3+</sup> doped materials, which are relevant to the visible laser operation, are compared in Table IV. Till now we have only observed a stimulated emission in the 790 nm band.

Comparison of the spectroscopic properties of Dy<sup>3+</sup> doped materials.

Comparison of the s	pectroscopic pro	operties of Dy <sup>3+</sup> d	oped materials.	
Host	$\Omega_2$	$\Omega_4$	$\Omega_6$	Refs.
	$[10^{-20} \text{ cm}^2]$	[10 <sup>-20</sup> cm	$1^{2}$ ] $[10^{-20} \text{ cm}^{2}]$	
LiNbO <sub>3</sub>	9.57	2.63	2.52	this work
LaF <sub>3</sub>	1.1	1.4	0.9	[14]
ZnO—Te0 <sub>2</sub> glass	4.28	1.32	2.53	[16]
ZnF <sub>2</sub> —CdF <sub>2</sub> glass	1.38	1.05	2.35	[17]
ZBLAN	2.7	1.8	2.0	[18]
Host	$^{4}F_{9/2}$ rad	$^4F_{9/2}$ fluor.	${}^{4}F_{9/2}$ - ${}^{6}H_{13/2}$	$^{4}F_{9/2}$ - $^{6}H_{15/2}$
	$\tau_{o}[\mu s]$	$ au_{A}[\mu s]$	$oldsymbol{eta_{ ext{means}}}$	$oldsymbol{eta}_{ ext{means}}$
LiNbO <sub>3</sub>	298	245	0.20	0.46
ZnO—TeO <sub>2</sub> glass	371	375	0.30	0.61
ZnF <sub>2</sub> —CdF <sub>2</sub> glass	2100	1260	0.27	0.5

#### 4. Conclusions

The spectroscopic properties of  $\mathrm{Dy}^{3+}$  ion in  $\mathrm{LiNbO_3}$  crystals were studied and analyzed. Polarized absorption spectra have been obtained leading to the determination of the energy level diagram and Judd—Ofelt intensity parameters. Because of the high effective phonon energy of the host matrix of 880 cm<sup>-1</sup>, dysprosium ion fluorescence in LNB is strongly affected by nonradiative decay, resulting in the absence of the  $^6H_{9/2}$  infrared emission. Visible  $\mathrm{Dy}^{3+}$  fluorescence presents similar lifetime to this of dysprosium doped oxide glasses, these observations are

supported by Cite theoretical calculations. Spectra registered at cryogenic temperatures indicated lower, than reported for other doped LiNbO<sub>3</sub> crystals, degree of the inhomogeneous line broadening and no multisite structure of lines was observed. Emission cross sections, which are relevant to the operation of a visible laser, have been determined and the pulsed infrared light generation at 750.5 nm has been observed for the first time in the plane parallel resonator. It is expected that optimization of the Dy<sup>3+</sup> ion concentration, resonator configuration and the crystal length will lead to low threshold operation at several other visible emission channels.

### References

- [1] L.F. Johnson, A.A. Ballman, J. Appl. Phys. 40, 297 (1969).
- [2] G. Dominiak-Dzik, S. Gołąb, I. Pracka, W. Ryba.-Romanowski, *Appl. Phys. A* 58, 551 (1994).
- [3] J. Garcia Sole, A. Monteil, G. Boulon, E. Camarillo, J.O. Tocho, I. Vergara, F. Jaque, *J. Phys. (France) JV1*, 403 (1991).
- [4] S.J. Field, D.C. Hanna, D.P. Scheperd, A.C. Tropper, Opt. Lett. 16, 481 (1991).
- [5] J.K. Jones, J.P. Dc Sandro, M. Hempstead, D.P. Sheppard, A.C. Large, A.C. Tropper, J.S. Wilkinson, *Rare Earth Doped Optical Waveguides, Les Houches, 22 June 1994*, Ed. B. Jacquier, University of Lyon, 1994, p. 827.
- [6] J. Sochtig, R. Gros, I. Bauman, W. Solder, H. Schutz, R. Widmer, *Electron. Lett.* 31, 551 (1995).
- [7] B.N. Samson, J.A. Medeiros Neto, R.I. Laming, D.W. Hewak, *Electron. Lett.* 30, 968 (1994).
- [8] B.N. Samson, J.A. Medeiros Neto, R.I. Laming, D.W. Hewak, *Electron. Lett.* 30, 1617 (1994).
- [9] L.F. Johnson, 11.1. Guggenheim, Appl. Plays. Lett. 23, 96 (1973).
- [10] D.M. Gill, J.C. Wright, L. McCaughan, Appl. Lett. 64, 2483 (1994).
- [11] B.R. Judd, *Plays. Rev.* 127, 750 (1962).
- [12] G.S. Ofelt, J. Chem. Phys. 37, 511 (1962).
- [13] A.A. Kaminskii, *Laser Crystals*, 2nd ed., Springer, Berlin 1990.
- [14] W.T. Carnall, H. Crosswhite, H. M. Crosswhite, *Energy Level Structure and Transit ion Probabilities of the Trivalent Lanthanides in LaF*<sub>3</sub>, Argonne National Laboratory, Argonne (Illinois) 1975.
- [15] C.A. Morrison, R.P. Leavit, in: *Handbook on the Physics and Chemistry of Rare-Earth*, Eds. K.A. Gschneidner, L. Eyring, North-Holland Publ.Co., New York 1982, p. 461.
- [16] J. Hormadaly, R. Reisfeld, J. Non-Cryst. Solids 30, 337 (1979).
- [17] R. Cases, M.A. Chamarro, R. Alcala, V.D. Rodriguez, J. Lumin. 48-49, 509 (1991).
- [18] V.M. Orera, P.1. Alonso, R. Cases, 11. Alcala, Phys. Chem. Glasses 20, 59 (1988).